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RESEARCH ARTICLE

SOME PROTECTIVE EFFECTS OF GINGER AGAINST CCI₄ INDUCED TOXICITY IN THE ADRENAL CORTEX OF ADULT WISTAR RATS (*Rattus norvegicus*)

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ARTICLE INFO	ABSTRACT	
Article History: Received 17 th December, 2012 Received in revised form 26 th January, 2013 Accepted 24 th February, 2013 Published online 19 th March, 2013	 Aims: To investigate some protective effects of ginger on CCl₄ (Carbon tetrachloride) induced toxicity in the adrenal cortex of adult wistar rats. Study design: Histological and Biochemical study. Place and Duration of study: Department of Anatomy, Faculty of Basic Medical Sciences, LAUTECH, Nigeria between September 2012 and December 2012. Methodology: Twenty-four adult healthy wistar rats of both sexes of average weight 210±4.22g were randomly 	
<i>Key words:</i> Adrenal cortex, Antioxidants, Carbon tetrachloride, Ginger, Wistar rats.	$-$ assigned into 4 groups four groups (N=6) such that Γ_1 , Γ_2 and Γ_3 served as treatment groups, while C served as the control group. T_1 received 2g of ginger and 2mls of CCl ₄ , T_2 were given 2mls of CCl ₄ Carbon tetrachloride while	
	T ₃ received 2g of ginger. The control group C was given distilled water. All the animals were exposed for 7 days. At the end of administration, all the rats were sacrificed cervical dislocation and processed immediately for histological techniques and bioassay of some antioxidant enzymes as well as lipid peroxidation. Results: Oxidative stress enzymes Superoxide dismutase, Glutathione reductase and Glutathione peroxidase as well as Glutathione levels were significantly (p<0.05) reduced in T ₂ compared to the control but relatively increased in T ₁ and T ₃ while Lipid peroxidation level was drastically reduced in T ₃ and to some extent in T ₁ compared to increased level in T ₂ . The histoarchitecture in the treatment group T ₃ and control C revealed distinct and normal pyramidal cells and freely anastomosing polyhedral cellular distribution in the cellular zones of adrenal cortex. Treatment group T ₁ also showed same normal histological presentation with few distortions. However the pictorial representation in treatment group T ₂ showed pyknotic pyramidal cells characterized with vacoulations specifically in the zonal fasciculata and to lesser extent in the zonal reticulosa. Conclusion: Ginger offers some ameliorative protections to the pyramidal and polyhedral cells of the adrenal cortex following CCl ₄ induced toxicity in wistar rats and also further affirms its antioxidative potentials	
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INTRODUCTION

Medicinal plants have continued providing valuable therapeutic agents, both in modern and in traditional medicine (Krentz et al, 2005). With the associated side effects of the modern medicine, traditional medicines are gaining importance and are now being studied to find the scientific basis of their therapeutic actions (Gupta and Briyal, 2004). Medicinal plants, their fractions and bioactive compounds play crucial role in detoxification of such toxins and scavenge free radicals (Sahreen et al., 2010). Ginger (Zingiber officinale) which belongs to the family Zingiberaceae is an example of botanicals that is vastly gaining popularity among modern physicians and its underground rhizomes are the medicinally useful part (Mascolo et al., 1989). Ginger is a strong anti-oxidant substance and may either mitigate or prevent generation of free radicals as its roots contain polyphenol compounds (6-gingerol and shogaols), which have a high antioxidant activity (Stoilova et al., 2007). It is considered a safe herbal medicine with only few and insignificant adverse/side effects (Ali et al., 2008). In addition, ginger reported as detoxifying agent against alcohol abuse (Shati and Elsaid, 2009) and bromobenzene intoxication (El-Sharaky et al, 2009). The adrenal gland is reported to be the most common endocrine organ associated with chemically induced lesions (Ribellin, 1984). The adrenal gland is exquisitely sensitive to toxic assault as it has been reported that the most frequently observed site of endocrine lesion is the adrenal gland (Ribelin, 1984 and Harvey, 1999). There are two features of the

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adrenal gland which make it vulnerable to toxic assault (Hinson and Raven, 1999). It is a discrete gland and its high vascularity, facilitates the delivery of toxins and metabolic substrates as well as the efficient removal of steroid products (Vinson and Hinson, 1992). Carbon tetrachloride (CCl₄) is known to induce damage in liver, lungs, kidneys, adrenals and central nervous system in humans and experimental animals (Rechnagel et al., 1989). Carbon tetrachloride (CCl4) is a potent hepatotoxic agent causing hepatic necrosis and nephrosis, and is widely used in animal models for induction of acute and chronic injury (Khan et al., 2010). It has been found that metabolism of CCl_4 involves the production of highly fatal trichloromethyl radical (CCl3•) and proxy trichloromethyl (•OOCCl3) free radicals through P450 bioactivation (Weber et al., 2003; Khan and Ahmed, 2009). CCl₄ is capable of causing lipid peroxidation and decreases activity of antioxidant enzymes (Adewole et al., 2007). Endogenous antioxidants such as polyphenolic compounds, ascorbic acid and monosaccharides in medicinal plants may constitute antioxidative defense by scavenging free radicals possibly increase the longevity of biological systems (Khan and Ahmed, 2009). This study was carried out to evaluate some protective effects of Ginger (Zingiber officinale) against CCl₄ induced adrenal toxicity.

MATERIALS AND METHODS

Twenty-four adult healthy wistar rats of both sexes of average weight 210±4.22g were maintained under standard laboratory conditions for an acclimatization period of 2 weeks in the animal holdings of

Anatomy Department, Ladoke Akintola University of Technology Ogbomoso. During which they were fed with standard laboratory mouse chow (Ladokun feeds, Ibadan) and provided water *ad libitum*. Daily weights were taken and documented. At the end of acclimatization, the rats were randomly assigned into four groups (N=6) such that T₁, T₂ and T₃ served as treatment groups, while C served as the control group. T₁ received 2g of ginger by oral administration and 2mls of Carbon tetrachloride, intraperitoneally, T₂ were given 2mls of Carbon tetrachloride, intraperitoneally while T₃ received 2g of ginger by oral administration. The control group C was given distilled water. All the animals were exposed for 7 days. At the end of administration, all the rats were sacrificed cervical dislocation and processed immediately for histological techniques and bioassay.

Enzyme Assay

For the measurement of the enzymes antioxidant activities, GSH and Lipid peroxidation, part of the adrenal gland specimens were weighed and homogenized in a sucrose buffer (0.25M Sucrose, 10Mm HEPES,1mMEDTA,pH 7.4) and the homogenate was centrifuged at 1000 x g for 60min at 4°C for the assay of superoxide dismutase (SOD). The activity of Superoxide dismutase (SOD) was measured by the method of Marklund et al, 1974 with some modification, an assay based on the ability of SOD to inhibit the autoxidation of pyrogallol by 50%. The assay mixture of 1 ml contained in final concentration, 50 mMSodium phosphate buffer (pH 7.4), 0.1 mM EDTA, 0.48 mMPyrogallol and appropriate amount of tissue extract containing 7-10 µg of protein. The change in absorbance of assay mixture was monitored spectrophotometrically at 420 nm for 3 min at 25°C against blank. One unit of enzyme activity is defined as the amount of enzyme that causes 50% maximal inhibition of pyrogallol autoxidation. The remaining brain tissues were homogenized in cold phosphate buffered saline (PBS) with 1mM EDTA and the homogenate was centrifuged at 1200 x g for 15 minutes at 4°C and the supernatant were analyzed for the activities of the following parameters:

Glutathione reductase (GR)

The Glutathione reductase (GR) activity was measured by the modified method of Erden *et al*, 1984. The reaction mixture of 1ml contained in final concentration, 4.1 mMTris-HCl (pH 7.5), 15mM MgCl2, 5.7 mM EDTA, 60 mMKCl, 2.6 mM Glutathione (oxidized) and 0.1 mM NADPH. The reaction was started by the addition of tissue extract containing approximately 100 μ g of protein. One unit of enzyme activity is defined as 1 μ mol of NADPH oxidized/min/mg protein. The decrease in absorbance was monitored at 25°C at 340 nm.

Glutathione peroxidase (GPx) which was determined using modified method as described by Lawrence and Burk.1976. The assay mixture of 1 ml contained in final concentration, 10 mM Potassium phosphate buffer (pH 7.0), 25 mM EDTA, 0.5mM Glutathione (reduced), 2mM Sodium azide, 1.5 IU Glutathione reductase, 0.1 mM NADPH and the cytosolic fraction containing about 50 μ g of protein. The reaction was started by the addition of t-butyl hydroperoxide and the decrease in absorbance was monitored at 25°C at 340 nm. One unit of enzyme activity is defined as 1 μ mol of NADPH oxidized/min/mg protein.

Glutathione (GSH) using commercially available kit (NWK-GSH01, North West Life science specialities, LLC., as previously described by Teitze, 1969.

Malanoaldehyde (MDA)

The level of lipid peroxidation was assessed in the adrenal gland tissue by measuring the formed malondialdehyde (MDA), an end product of fatty acid peroxidation, using thiobarbituric acid reactive substance(TBARS) method (Genet, *et al*, 2002).10% tissue homogenate was centrifuged at 1000xg for 10 min and deproteinized with half volume of 20% trichloroacetic acid (TCA). The supernatant in 10 mM Potassium phosphate buffer (pH 7.4) was incubated at 80°C

for 15 min in water bath with 0.53% Thiobarbituric acid in glacial acetic acid and centrifuged. The concentration of MDA-TBA complex was determined spectrophotometrically at 532nm against blank.

Statistical analysis

The data were analyzed using the computerized statistical package 'SPSS Version 11'. Mean and standard error of mean (SEM) values for each experiment group was determined. The means were compared by analysis of variance at a level of significance of 95%.

RESULTS

Biochemical quantification

Biochemical quantification revealed a slightly decreased values in the specific activities of SOD in the treatment group T2 with Mean±SEM (0.48±0.28)U/mg protein compared to the control group C with Mean±SEM (1.71±1.1)U/mg protein while group T1 and T3 recorded significantly (P < 0.05) increased values Mean±SEM (1.82±1.32 and 1.88±1.22)U/mg protein respectively. The results obtained for GR showed that the specific activity was drastically reduced in group T2 with Mean±SEM (11.49±0.27)*mU/mg protein* compared to the control group C with Mean±SEM(18.55±4.03)mU/mg protein but on the contrast the groups T1 and T3 recorded significantly (P < 0.05) increased values Mean±SEM (27.01±2.81 and 31.80±3.11)mU/mg protein respectively while the specific activity of GPx was decreased significantly(P<0.05) for treatment group T2 with Mean±SEM (0.38±0.26)mU/100mg protein compared to the control group C with Mean±SEM(1.79±1.39)mU/100mg protein while group T1and T3 recorded significantly (P < 0.05) increased values Mean±SEM (3.59±0.92 and 3.82±0.71)mU/100mg protein respectively as shown in Table 1.

 Table 1. Mean ± SEM of Antioxidant Enzyme Activities in the adrenal gland

GROUPS	SOD(U/mg protein)	GPxmU/100mg protein	GRmU/mg protein
С	1.71±1.10	1.79±1.39	18.55±4.03
T1	1.82±1.33	3.59±0.92	27.01±2.81
T2	0.48 ± 0.28	0.38±0.26	11.49±0.27
T3	1.88 ± 1.22	3.82±0.71	31.80±3.11

Significantly reduced values (P<0.05) was recorded in the GSH levels of Mean±SEM (0.86±0.13) $\mu M/mg$ protein in groups T2 when compared to Mean±SEM (2.72±1.29) $\mu M/mg$ protein obtained in the control group while group T1 and T3 recorded significantly (P < 0.05) increased values Mean±SEM (3.19±1.09 and 4.15±1.23) $\mu M/mg$ protein respectively while MDA level was drastically reduced in T3 with Mean±SEM (0.62±0.22) $\mu M/mg$ protein and to some extent in T1 with with Mean±SEM (0.95±0.82) $\mu M/mg$ protein compared to increased level in T2 with Mean±SEM (2.37±0.46) $\mu M/mg$ protein while control group value stood at Mean±SEM (1.51±0.11) $\mu M/mg$ protein as shown in Table 2.

Table 2. (Mean \pm SEM) $\mu M/mg$ protein of GSH and LPO Levels in the adrenal gland

GSH	LPO
2.72±1.29	1.51±0.11
3.19±1.09	0.95±0.82
0.86±0.13	2.37±0.46
4.15±1.23	0.62 ± 0.22
	GSH 2.72±1.29 3.19±1.09 0.86±0.13 4.15±1.23

Histological Findings

The histoarchitecture in the treatment group T3 and control C revealed distinct and seemingly normal cellular distribution of cellular zones of zona grannulosa (ZG), zona fasciculata (ZF) and zona reticulosa (ZR) with no histoarchitectural distortions. Treatment group T_1 also showed same normal histological presentation with few distortions. However

the pictorial representation in treatment group T2 showed pyknotic polyhedral cells with vacoulations in the zona fasciculata and to lesser extent in the zonal reticulosa.

DISCUSSION

There is increasing evidence that various chemicals introduced into the environment have the potential to disrupt the endocrine system in humans and wildlife (Waslat and Hanaa, 2011).



Figure 1: Photomicrograph of adrenal cortex (control section C) showing the different cellular zones of zona grannulosa (ZG), zona fasciculata (ZF) and zona reticulosa (ZR). Note the blue arrow pointing to the connective tissue capsule as well as the Black arrows pointing to the normal and well distributed pyramidal cells in ZG, polyhedral cells in ZF and ZR. H&E stain x400.



Figure 2. Photomicrograph of adrenal cortex (Treatment section T_1) showing the different cellular zones of zona grannulosa (ZG), zona fasciculata (ZF) and zona reticulosa (ZR). Note the orange arrow pointing to the adipose tissue surrounding the adrenal gland while the blue arrow is pointing to the connective tissue capsule .The Black arrows point to the normal and well distributed pyramidal and polyhedral cells while the green arrows show the few distorted cells. H&E stain x400.

As clearly evident from results obtained in this study, Carbon tetrachloride (CCl₄) induced alterations in the histoarchitecture, antioxidant enzymes and lipid peroxidation which thus clearly suggest that CCl₄ is able to cause oxidative stress in adrenal gland. The histological presentations in this study as evident from Figure 3 revealed distorted and seemingly pyknotic cells with vacoulations in the zona fasciculata and the effects were less pronounced in zona reticulosa and zona glomerulosa which could have been due to damaging effects of CCl₄. However Figures 1 and 4 revealed normal histoarchitecture while the effects were less pronounced in Figure 2 and this is most likely due to a protective action of Ginger. Reports have it that the most frequently implicated lesion site of adrenal cortex is the zona fasciculate (Rosol *et al*, 2001).



Figure 3. Photomicrograph of adrenal cortex (Treatment section T_2) showing the different cellular zones of zona grannulosa (ZG), zona fasciculata (ZF) and zona reticulosa (ZR). Note the blue arrow pointing to the connective tissue capsule .The Black arrows point to the distorted polyhedral cells in the zona fasciculata and zona reticulosa with vacoulation (V) sites .H&E stain x400.

The distorted cells and vacoulations seen in this study is in conformation with the earlier findings of Sakr, et al, 2002 where atrophy and vacoulation of the zona fasciculate was observed following exposure to methylpredinysolone. The histological findings seen here revealed a seemingly protective potential of ginger which is ameliorating the damaging effects of CCl₄ in the ginger exposed group. Determination of activity level of these antioxidant enzymes is an appropriate indirect way to assess the pro-oxidant antioxidant status in tissues (Priscilla and Prince, 2009). It is a known fact that generation of highly ROS such as superoxide radicals, hydrogenperoxide, hydroxl radicals and LPO in the presence of heavy metals ions are known to damage various cellular components including proteins, membrane lipids and nucleic acids (Halliwell and Gutheridge, 1989) as both enzymatic and nonenzymatic antioxidants exist in the intracellular and extracellular environment to detoxify ROS (Frie et al, 1988). Antioxidants act as radical scavenger, hydrogen donor, electron donor, peroxide decomposer, singlet oxygen quencher, enzyme inhibitor, synergist, and metal-chelating agents. In this study, determined activities of the antioxidant defense system like SOD, GR and GPx showed decreased values in the CCl₄ exposed group compared to the control but higher values were recorded in the Ginger exposed group as seen in Table 1 and this is also in conformity with earlier findings of Moreira et al.2001. Superoxide dismutases are specific for catalytic removal of superoxides by converting them into

H₂O₂ (Halliwell and Gutteridge, 2007). The decreased activity of SOD may be a response to increased production of H_2O_2 and O_2 as well as decreased protein expression levels as reported by Argano et al. 1997. The decreased activity of enzymes could also be due to their decreased protein expression levels from CCl₄ toxicity condition as reported earlier by Sindhu et al, 2004. An antagonistic effect between selenium (as a cofactor) and CCl₄ may affect GPx activity and it can thus render GPx a potential target for CCl₄ toxicity because a reduction in selenium uptake may increase the susceptability of cell to oxidative stress. It is clear that GPx needs GSH to decompose H₂O₂ or other peroxides with the simultaneous oxidation of GSH into GSSG, however, GR which is another component of the antioxidant defense system will reduce GSSG back to GSH and thereby will support the antioxidant defense mechanism indirectly. The presence of disulfide at the active site of GR as earlier reported by [Quinlan, 1988], may be a target for CCl₄ and will result in inhibition of GR activity. GSH is a predominant endogenous antioxidant and used as a cofactor to remove hydrogen peroxide and lipoperoxides by the GSH-Px family during which GSH is converted into oxidized form of glutathione (GSSG) (Muhammad and Tahira, 2011). In this study as seen from Table 2, significantly reduced level of GSH was recorded in the CCl₄ exposed group T2 compared to the control while the Ginger exposed groups T1 and T3 recorded values at par with the control and this clearly portrayed Ginger as an ameliorative oxidant. Oxidized glutathione is converted back into GSH by another rate controlling enzyme the glutathione reductase (GSR) thereby maintain the intracellular GSH levels. This optimum level of GSH is an utmost criterion in maintaining the structural integrity and physiology of cell membranes.



Figure 4. Photomicrograph of adrenal cortex (Treatment section T_3) showing the different cellular zones of zona grannulosa (ZG), zona fasciculata (ZF) and zona reticulosa (ZR). Note the blue arrow pointing to the connective tissue capsule as well as the Brown arrows pointing to the normal and well distributed pyramidal cells in ZG, while the Black arrows point to the normal polyhedral cells in ZF and ZR. H&E stain x400.

Lipid peroxidation is a free radical process involving a source of secondary free radical, which further can act as second messenger or can directly react with other biomolecules, enhancing biochemical lesions. Lipid peroxidation occurs on polysaturated fatty acid located on the cell membranes and it further proceeds with radical chain reaction. Hydroxyl radical initiates ROS and removes hydrogen atom, thus producing lipid radical and further converted into diene conjugate. Due to lipid peroxidation, a number of compounds are formed, for example, alkanes, malonaldehyde (MDA), and isoprotanes. When compared to the control and other groups in this study, significantly P<0.05 high level of MDA was recorded with extremely low values obtained in the Ginger exposed groups. Reactive oxygen species react with lipids and cause peroxidative changes that result in elevated lipid peroxidation. The increased lipid peroxidation with CCl₄ may be an indication of a decrease in non-enzymatic antioxidant of defense mechanism (Shanmugam et al, 2010). The MDA, another end product of lipid peroxidation has been demonstrated to be a mutagenic and genotoxic agent that can contribute to the development of human cancers (Feron et al, 1991). Hence, agents that can inhibit lipid peroxidation in organs with variations in the level of polyunsaturated fatty acids (PUFA) and antioxidants status will be an addition to the concept of functional food. Although previous studies have demonstrated the protective effect of Z. officinale against anticancer drugs induced organ damages in experimental animals (Ajith et al, 2007, 2008), the findings obtained from this study has also added to the available informations on the antioxidative potentials of ginger

REFERENCES

- Adewole SO, Salako AA, Doherty OW, Naicker T (2007). Effect of melatonin on carbon tetrachloride-induced kidney injury in Wistar rats. Afr. J. Biomed. Res., 10: 153-164.
- Ajith TA, Aswathi S, Hema U (2008). Protective effect of *Zingiber* officinale roscoe against anticancer drug doxorubicin-induced acute nephrotoxicity. Food Chem Toxicol; 46: 3178-81
- Ajith TA, Hema U, Aswathi S (2007). Zingiber officinale Roscoe prevents acetaminophen- induced acute hepatotoxicity by enhancing hepatic antioxidant status. Food Chem Toxicol; 45: 2267-72.
- Ali SS, Kasoju N, Luthra A, Singh A, Sharanabasava H, Sahu A,Bora U (2008). Indian medicinal herbs as sources of antioxidants. Food Res. Int., 41: 1-15
- aluminum (III) ions on iron stimulated lipid peroxidation in liposomes, erythrocytes and rat liver microsomal fractions. Biochem. Biophys. Acta. 962:196-200.
- Argano M, Brignardello E, Tamango O, Bocuzzi G (1997). Dehydroeppiandrosterone administration prevents the oxidative damage induced by acute hyperglycemia in rats. J Endocrinol;155:233-240.
- El-Sharaky, A. S., Newairy, A. A., Kamel, M. A. and Eweda, S.M. (2009). Protective effect of ginger extract against bromobenzene-induced hepatotoxicity in male rats. *Food Chem. Toxicol.* 47(7): 1584–1590
- Erden M, Bor NM (1984). Changes in reduced glutathione, glutathione reductase and glutathione peroxidase after radiation in guinea pigs. Biochem ;31:217-227
- Feron VJ, Til HP, de Vrijer F, Woutersen RA, Cassee FR, van Bladeren PJ. (1991).Aldehydes: occurrence, carcinogenic potential,mechanism of action and risk assessment. Mutat Res ; 259:363-85.
- Frie B, Stocker R (1988) Ames BN. Antioxidant defences and lipid peroxidation in human blood plasma. Proc Natl Acad Sci. 37:569–71
- Genet S, Kale RK, Baquer NZ (2002). Alteration in antioxidant enzymes and oxidative damage in experimental diabetic rat tissues: effect of vanadate and fenugreek (*Trigonella foenum* graecum). Mol Cell Biochem. ;236:7-12.
- Gupta YK, Briyal S (2004). Animal models of cerebral ischemia for evaluation of drugs. Indian J PhysiolPharmacol., 48: 379–94.
- Halliwell B and Gutteridge JMC (2007). Cellular responses to oxidative stress: adaptation, damage, repair, senescence and death. In: Free radicals in biology and medicine, Oxford University Press Inc., Oxford, pp.187-267.
- Halliwell B, Gutheridge JMC (1989). Protection against oxidants in biological systems: The superoxide theory of oxygen toxicity. In. Free radical in biology and medicine Clarendon Press,Oxford. ; 86-123.
- Harvey PW (1999). An overview of adrenal gland involvement in toxicology. *In*: Harvey PW (ed.). The Adrenal in Toxicology. Taylor and Francis, London, pp.3-19.

- Hinson JP and Raven PW (1999). Adrenal toxicolog. *In*: Harvey PW, Rush KC and Cockburn A (eds.) Endocrine and Hormonal Toxicology. Chichester, Wiley, pp.67-90.
- Khan MR, Ahmed D (2009). Protective effects of *Digera muricata* (L.) Mart. On testis against oxidative stress of carbon tetrachloride in rat. Food Chem. Toxicol., 47: 1393-1399.
- Khan RA, Khan MR, Sahreen S, Bokhari J (2010). Prevention of CCl4-induced nephrotoxicity with *Sonchus arvensis*in rat. Food Chem.Toxicol., doi:10.1016/j.fct.2010.06.016.
- Krentz AJ, Bailey CJ (2005). Oral antidiabetic agents: current role in type 2 diabetes mellitus. Drugs, 65: 385 -411.
- Lawrence RA, Burk RF (1976). Gutathione peroxidase activity in selenium deficient rat liver.Biochem Biophys Res Commun. ;71:1161-1165.
- Maklund S, Maklund G (1974). Involvement of superoxide anion radical in the autoxidation of pyrogallol and a convenient assay for superoxide dismutase. Eur J Bichem ;47:469-474
- Mascolo NR, Jain SC, Jain FC (1989). Ethnopharmacologic investigation of ginger (*Zingiber officinale*). J. Ethnopharmacol., 27: 129-140.
- Moreira EG, Rossa GJ, Barros SB, Vassilieff VS, Vassilieff I (2001) Antioxidant defens in rat brain regions after developmental lead exposure. Toxicology. 169:145-151.
- Priscilla DH and Prince PSM (2009). Cardioprotective effect of gallic acid on cardiac troponin-T, cardiac marker enzymes, lipid peroxidation products and antioxidants in experimentally induced myocardial infarction in Wistar rats. *Chem. Biol. Inter.*, 179: 118-124.
- Quinlan GJ, Halliwell B, Moorhouse CP, Gutteridge JMC (1988). Action of lead (II) and
- Rechnagel RO, Glende EA, Dolak JA and Waller RL (1989). Mechanisms of carbon tetrachloride toxicity. *J.Pharmacol. Exper. Therap.*, 43: 139-154.
- Ribelin WE (1984). The effects of drugs and chemicals upon the structure of the adrenal gland. *Fundam Appl Toxicol*, 4:105–119.
- Rosol TJ, Yarrington JT, Latendresse J. and Capen CC.(2001). Adrenal gland: structure, function, and mechanisms of toxicity. Toxicol Pathol., 29: 41-48

- Sahreen S, Khan MR, Khan RA (2010). Evaluation of antioxidant activities of various solvent extracts of *Carissa opaca* fruits, 22: 1205-1211.
- Sakr SA,El-Desouky NI and Hanafy SM(2002). Methylprednisolone induced histological and histochemical changes in adrenal cortex of rats. Online journal of biological sciences ;2(5):320-324
- Shanmugam KR, Ramakrishna CH, Mallikarjuna K, Sathyavelu Reddy (2010).Protective effects of ginger against alcoholinduced renal damage and antioxidant enzymes in male albino rats. Indian journal of experimental biology,48:143-149
- Shanmugam KR,Ramakrishna CH,Mallikarjuna K, Sathyarvelu KR (2010).Protective effects of ginger against alcohol-induced renal damage and antioxidant enzymes in male albino rats. Indian journal of experimental biology. 48:143-149
- Shati AA, Elsaid FG (2009). Effects of water extracts of thyme (*Thymus vulgaris*) and ginger (*Zingiber officinale* Roscoe) on alcohol abuse. Food. Chem. Toxicol., 47: 1945-1949.
- Sindhu RK, Koo JR, Roberts CK, Vaziri ND (2004). Dysregulation of hepatic superoxide dismutase, catalase and glutathione peroxidase in diabetes response to insulin and antioxidant therapies. Clin Exp Hypertens; 26:43-53.
- Stoilova I, Krastanov A, Stoyanova A, Denev P, Gargova S (2007). Antioxidant activity of a ginger extract (*Zingiber* officinale). Food Chem., 102: 764-770
- Teitze F (1969). Enzymatic method for the quantitative determination of nanogram amount of total and oxidized glutathione. Application to mammalian blood and other tissues. Annal Biochem. ;27:502-522
- Vinson GP and Hinson JP (1992). Blood Flow and Hormone Secretion in the Adrenal gland. In: James VHT (ed.). The Adrenal Gland, Raven Press, New York, pp.71-86
- Waslat W. Elshennawy and Hanaa R. Aboelwafa (2011). Structural and ultrastructural alterations in mammalian adrenal cortex under influence of steroidogenesis inhibitor drug. Journal of American Science;7(8)
- Weber LW, Boll M, Stampfl M (2003). Hepatotoxicity and mechanism of action of haloalkanes: carbon tetrachloride as a toxicological model. Crit. Rev. Toxicol., 33: 105-136.
